

# **Safety Data Sheet**

**Revision Number: 4** 

#### 1. Identification of the Substance/Preparation and of the Company/Undertaking

Revision date 31-Dec-2018

Product identifier Product Description:	Magnaphos Plates / Magnaphos Strips
<u>Other means of identification</u> Product code UN/ID no. Registration number(s)	12-MAG2 UN2011 70506-309
<u>Recommended use of the chemical</u> Recommended use Uses advised against	and restrictions on use Restricted Use Pesticide. Fumigant. Activities contrary to label recommendation
Details of the Supplier of the Safety Supplier Address UPL NA Inc. 630 Freedom Business Center Suite 402 King of Prussia, PA 19406	<u>v Data Sheet</u>
Emergency telephone number Company Phone Number Emergency telephone number	1-800-438-6071 Chemtrec: (800) 424-9300 (24hrs) or (703) 527-3887 Medical: Rocky Mountain Poison Control Center (866) 673-6671 (24hrs)

2. Hazards Identification

**Classification** 

#### **OSHA Regulatory Status**

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200)

Acute toxicity - Oral	Category 2
Acute toxicity - Inhalation (Gases)	Category 1
Acute toxicity - Inhalation (Vapors)	Category 1
Substances or mixtures which, in contact with water, emit flammable gases	Category 1

#### Label elements

#### EMERGENCY OVERVIEW

# DANGER

Hazard Statements Fatal if inhaled FATAL IF SWALLOWED Harmful in contact with skin Very toxic to aquatic life In contact with water releases flammable gases which may ignite spontaneously



Precautionary Statements - Prevention

Do not eat, drink or smoke when using this product Do not handle until all safety precautions have been read and understood Protect from moisture Wear eye/face protection Wear protective gloves Wash hands thoroughly after handling

IF INHALED Immediately call a POISON CENTER or doctor/physician IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician Rinse mouth

#### Precautionary Statements - Storage

Store locked up Store in a well-ventilated place. Keep container tightly closed

#### **Precautionary Statements - Disposal**

Refer to manufacturer/supplier for information on recovery/recycling

# Hazards Not Otherwise Classified (HNOC)

OTHER INFORMATION

Very toxic to aquatic life

May be harmful in contact with skin

### 3. Composition/information on Ingredients

Chemical name	CAS No	Weight-%
Magnesium phosphide	12057-74-8	56.0 (a.i.)

If CAS number is "proprietary", the specific chemical identity and percentage of composition has been withheld as a trade secret.

4. First aid measures		
FIRST AID MEASURES		
Eye contact	Hold eye open and rinse slowly and gently with water for 15 - 20 minutes. Remove contact lenses, if present, after 5 minutes, then continue rinsing eye. Call a poison control center or doctor for treatment advice.	
Skin contact	Take off contaminated clothing. Wash off immediately with plenty of water for at least 15 minutes. Call a poison control center or doctor for treatment advice.	
Inhalation	Move person to fresh air. If person is not breathing, call 911 or an ambulance, then give	

	artificial respiration, preferably mouth-to-mouth if possible. Call a physician or poison control center immediately.
Ingestion	Call a physician or poison control center immediately. Do not induce vomiting without medical advice. Never give anything by mouth to an unconscious person. Vomiting may off-gas and release phosphine, which could pose a risk of secondary contamination.
Protection of First-aiders	Use personal protective equipment.
Most Important Symptoms and Effe	cts, Both Acute and Delayed
Most Important Symptoms and Effects	Headache. Diarrhea. Nausea. Difficulty in breathing. Dizziness.
Indication of Any Immediate Medica	I Attention and Special Treatment Needed
Notes to physician	<ul> <li>Magnesium phosphide- This material reacts with moisture from air, water, acids, and manyother liquids to release hydrogen phopshide (phosphine) gas. Symptoms of severe poisoning may occur within a few hours to several days. Phosphine poisoning may result in pulmnary edema, liver elevated serum, GOT, LDH and alkaline phosphatase reduced prothrombin, hermorahage and jaundice and kidmey hematuria. Pathology is characterized by hypoxia.</li> <li>Mild inhalation exposure causes malaise, ringing of ears, fatigue, nausea, and pressure in the chest, which is relieved by removal to fresh air. Moderate poisoning causes weakness, vomiting, and pain just above the stomach, chest pain, diarrhea and dyspnea. Symptoms of severe poisoning may occur within a few hours to several days, resulting in pulmonary edema and may lead to dizziness, cyanosis, unconsciousness and death.</li> <li>In sufficient quantity, phosphine affects the liver, kidneys, lungs, nervous system, and circulatory system. Inhalation can cause lung edema, and hyperemia. Ingestion can cause lung and brain symptoms but damage to the viscera is more common. Phosphine poisoning may result in (1) pulmonary edema, (2) liver elevated serum GOT, LDH and alkaline phosphatase, reduced prothrombin, hemorthage and jaundice and (3) kidney hematuria and anuria. Pathology is characterized by hypoxia. Trequent exposure to subacute concentrations over a period of days or weeks may cause poisoning. Treatment is symptomatic.</li> <li>In tis milder forms, symptoms of poisoning may take some time, up to 24 hours, to make their appearance and the following is suggested:</li> <li>1) Give complete rest for 1 -2 days during which the patient must be kept quiet and warm.</li> <li>2) Should patient suffer from vomiting or increased blood sugar, appropriate solution should be administration of cardiac and circulatory stimulants.</li> <li>In cases of severe poisoning (Intensive Care Unit recommended) :</li> <li>1) Where pulmonary edema, venesection should be performed under vein</li></ul>

# 5. Fire-fighting measures

<u>Suitable extinguishing media</u> Carbon dioxide (CO2). Use:. Dry chemical. alcohol-resistant foam.

Magnesium phosphide is not flammable; however, it reacts with water to produce hydrogen phosphide (phosphine) gas which may ignite spontaneously at concentrations above the LEL of 1.8% v/v.

#### Unsuitable extinguishing media Aquatic. Water spray.

#### Specific hazards arising from the chemical

Thermal decomposition can lead to release of irritating gases and vapours.

Metal phosphides: Hydrogen phosphide (phosphine)/air mixtures at concentrations above the lower flammable limit may ignite spontaneously. Ignition of high concentrations of hydrogen phosphide can produce a very energetic reaction. Explosions can occur under these conditions and may cause personal injury. NEVER allow build up of hydrogen phosphide to exceed explosive concentrations. Containers of metal phosphides should be opened in open air and never in a flammable atmosphere. Do not confine spent or partially spent dust as slow release of hydrogen phosphide are piled in contact with liquid water. Fires containing metal phosphides or hydrogen phosphide will produce phosphoric acid by the following reaction: 2PH3 + 4O2 = H2O + P2O5 = 2H3PO4.

#### Hazardous combustion products Phosphine gas.

#### Explosion data

#### Protective equipment and precautions for firefighters

Use personal protective equipment. As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

#### 6. Accidental release measures

#### Personal precautions, protective equipment and emergency procedures

Personal Precautions	Avoid contact with skin and eyes. An accidental spill/release of material may produce high levels of gas. A NIOSH/MSHA approved full face gas mask with phosphine cartridge of SCBA must be employed during wet deactivation of partially spent material. Wear protective gloves and clothing. Wear protective gloves/protective clothing and eye/face protection.
Environmental Precautions	
Environmental precautions	Consult a regulatory specialist to determine appropriate state or local reporting requirements, for assistance in waste characterization and/or hazardous waste disposal and other requirements listed in pertinent environmental permits.
Methods and material for containme	ent and cleaning up
Methods for Clean-Up	Damaged plates should be transferred, only by individuals who are knowledgeable of magnesium phosphide properties and who employ protective gloves and appropriate respiratory protection, to a sound metal container for transfer to a secure location.

#### 7. Handling and Storage

#### Precautions for safe handling

HandlingDo not eat, drink or smoke when using this product. Keep out of reach of children. Read<br/>label, manual and safety data sheet BEFORE handling. Always employ protective gloves,<br/>clothing and appropriate respiratory protection. It is recommended that product be opened<br/>in air or near a fan that exhausts outside immediately. Never open in a flammable<br/>atmosphere to avoid, although rare, flash. When opening point away form face and body.<br/>Do not expose to atmospheric moisture any longer than necessary.<br/>Use of this product is STRICTLY PROHIBITED on single and multifamily residential<br/>properties and nursing homes, schools (except athletic fields) daycare facilities and<br/>hospitals.For burrowing rodent application: The use of this product is strictly prohibited within 100 feet<br/>of any building where humans and/or domestic animals do or may reside, on single or multi<br/>family residential properties and nursing homes, schools (except athletic fields) , day care<br/>facilities, and hospitals.

#### Conditions for safe storage, including any incompatibilities

Storage	Store in a secure area. Keep out of the reach of children. Keep containers tightly closed in a cool, well-ventilated place. Keep in properly labelled containers. Keep away from heat. Store in accordance with the particular national regulations. Store in accordance with local regulations.
incompatible materials	Aquatic. Hydrogen phosphide may react with certain metals (gold, silver, brass, other precious metals and their alloys) and cause corrosion especially at high temperatures and relative humdities. Small electric detectors, brass sprinkler heads, batteries and battery chargers, forklifts, temperature monitoring systems, electrical switch gear, communication devices, computers, calculators, watches and other electronic equipments shoul dbe protected or removed before fumigation. Hydrogen phosphide gas will also react with certain metallic salts and, therefore such items as photographic film, copying papers and some inorganic pigments, etc. should not be exposed.
	8. Exposure Controls/Personal Protection

Exposure guidelines	Guidelines for phosphine gas.
Engineering controls	Measurements of the concentration of Magnesium phosphide in the air must be provided and used to verify the concentration in the atmosphere.
Personal protective equipment	
Eye/Face Protection	Use eye protection to avoid eye contact. Where there is potential for eye contact have eye flushing equipment available. Safety glasses with side-shields.
Skin protection	Impervious butyl rubber gloves. Wear protective gloves/clothing. Socks and footwear.
Respiratory protection	When workers are facing concentrations above the exposure limit they must use appropriate certified respirators. In case of inadequate ventilation wear respiratory protection. A NIOSH approved air-purifying full face gas mask with a chin style mounted canister approved for phosphine may be employed for concentrations up to 15 ppm. At concentrations above that level, or when concentration is unknown, NIOSH/MESA approved SCBA or equivalent must be worn.

#### General hygiene considerations

Do not eat, drink or smoke when using this product. Wear respiratory protection. Wear suitable gloves and eye/face protection. Remove and wash contaminated clothing before re-use. Wash hands before breaks and immediately after handling the product.

# 9. Physical and Chemical Properties

#### Information on basic physical and chemical properties

Physical state Appearance	solid Plate Plate	Odor	Pure phosphine gas is odorless but a garlic odor might be detected due to a contaminant. Since odor may not be detected under certain circumstances, the absence of a garlic odor does not mean that phosphine gas is absent.
color	dark Gray		
<u>Property</u> pH Melting point/freezing point Boiling Point/Range Flash Point Evaporation Rate Flammability (solid, gas)	VALUES No information available no data available No information available No information available No information available Emission of phosphine (PH3), toxic	<u>Remarks/ • Method</u>	

gas extremely flammable in contact with water.

Flammability limit in air	
Upper Flammability Limit	No information available
Lower Flammability Limit	No information available
vapor pressure	No information available
Vapor Density	No information available
Specific gravity	No information available
Water solubility	No information available
Solubility in Other Solvents	No information available
Partition coefficient: n-octanol/wate	rNo information available
Autoignition temperature	no data available
Decomposition temperature	No information available
Viscosity, kinematic	No information available
Dynamic viscosity	No information available
Explosive properties	No information available
Oxidizing properties	No information available

#### **OTHER INFORMATION**

Softening point molecular weight VOC Content Liquid Density Bulk density No information available No information available No information available No information available 0.61 mg/L

#### **10. Stability and Reactivity**

# Reactivity

Water reactive

#### **Chemical stability**

Stable under normal conditions. Reacts with water to form hydrogen phosphide (phosphine) gas.

#### Possibility of hazardous reactions

None under normal processing.

#### Hazardous polymerization Hazardous polymerisation does not occur.

#### Conditions to avoid

Protect from moisture. Elevated temperatures, moisture and oxygen break down the product and induce flammable and toxic gas.

#### incompatible materials

Aquatic. Hydrogen phosphide may react with certain metals (gold, silver, brass, other precious metals and their alloys) and cause corrosion especially at high temperatures and relative humdities. Small electric detectors, brass sprinkler heads, batteries and battery chargers, forklifts, temperature monitoring systems, electrical switch gear, communication devices, computers, calculators, watches and other electronic equipments shoul dbe protected or removed before fumigation. Hydrogen phosphide gas will also react with certain metallic salts and, therefore such items as photographic film, copying papers and some inorganic pigments, etc. should not be exposed.

#### Hazardous decomposition products

Phosphine gas. Oxides of phosphorous. Oxides of carbon and nitrogen.

#### **11. Toxicological Information**

#### Information on Likely Routes of Exposure

#### Product information

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	Magnaphos plate: Acute oral LD50 (rat) = >5-50 mg/kg Acute dermal LD50 (rat) = 2633.05 mg/kg Acute inhalation LC50 (rat) = 50 ppm (1 hour) Eye irritation = Irreversible corneal opacity Skin sensitization = Not a sensitizer
Inhalation	Poison - may be fatal if inhaled.
Eye contact	Avoid contact with eyes. Contact with eyes may cause irritation.
Skin contact	May be absorbed through the skin in harmful amounts.
Ingestion	FATAL IF SWALLOWED.

#### Information on Toxicological Effects

#### Symptoms

No information available.

#### Delayed and immediate effects as well as chronic effects from short and long-term exposure\_

Sensitization	No information available.
Mutagenic effects	no data available.
Carcinogenicity	There are no known carcinogenic chemicals in this product.
Reproductive effects	Not Available.
STOT - Single Exposure	no data available.
STOT - Repeated Exposure	no data available.
Chronic toxicity	Avoid repeated exposure.
Target organ effects	Respiratory System, EYES, skin.
Aspiration hazard	No information available.

#### Numerical Measures of Toxicity - Product information

LD50 Oral	< 50 mg/kg (rat)
LD50 Dermal	2633.5 mg/kg (rat)
LC50 Inhalation	Inhalation LC50 50 ppm (1 hr)

#### **12. Ecological Information**

#### ecotoxicity

Highly toxic to wildlife

#### Persistence/Degradability

no data available.

#### **Bioaccumulation/ Accumulation**

Does not bioaccumulate.

#### **Other Adverse Effects**

no data available

# 13. Disposal Considerations

#### Waste Treatment Methods

Waste Disposal Method

Pesticide wastes are acutely hazardous. Improper disposal of excess pesticide or rinsate is a violation of Federal law. If the wastes cannot be disposed of by use or according to label

instructions, contact your State Pesticide or Environmental Control Agency, or the Hazardous Waste representative at the nearest EPA Regional Office for guidance. Follow label for proper disposal instructions.

#### **Contaminated packaging**

Refer to product label.

#### **14. Transport Information**

DOT	When shipped in bulk or internationally the marine pollutant marking must also be added to the package.
UN/ID no.	UN2011
Proper shipping name	Magnesium phosphide
Hazard class	4.3
Subsidiary class	(6.1)
Packing group	PG I
IATA UN/ID no. Proper shipping name Hazard class Subsidiary class Packing group Description	UN2011 Magnesium phosphide 4.3 (6.1) PG I
IMDG	Forbidden by passenger aircraft
UN/ID no.	UN2011
Proper shipping name	Magnesium phosphide
Hazard class	4.3
Subsidiary class	(6.1)
Packing group	PG I
Environmental hazards	IMDG - Marine Pollutant

### **15. Regulatory Information**

This chemical is a pesticide product registered by the Environmental Protection Agency and is subject to certain labeling requirements under federal pesticide law. These requirements differ from the classification criteria and hazard information required for safety data sheets, and for workplace labels of non-pesticide chemicals. Following is the hazard information as required on the pesticide label:

**Ventilation Control** 

Measurements of the concentration of magnesium phosphide must be provided and used to verify the concentration in the atmosphere.

Restricted Use Pesticide. Due to inhalation toxicity of phosphine gas. Keep out of Reach of Children. For burrowing rodent application: The use of this product is strictly prohibited within 100 feet of any building where humans and/or domestic animals do or may reside, on single or multi family residential properties and nursing homes, schools (except athletic fields), day care facilities, and hospitals. May be fatal if swallowed. Toxic to wildlife.

International Inventories	
USINV	Present
DSL/NDSL	Present
EINECS/	Not Present

ELINCS	
ENCS	Not Present
China	Not Present
KECL	Present
PICCS	Present
AICS	Present
TSCA	Present

 TSCA - United States Toxic Substances Control Act Section 8(b) Inventory

 DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

 EINECS/ELINCS - European Inventory of Existing Commercial Chemical Substances/EU List of Notified Chemical Substances

 ENCS - Japan Existing and New Chemical Substances

 IECSC - China Inventory of Existing Chemical Substances

 KECL - Korean Existing and Evaluated Chemical Substances

PICCS - Philippines Inventory of Chemicals and Chemical Substances

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AICS - Australian Inventory of Chemical Substances

#### Federal Regulations

#### SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product contains a chemical or chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372:

#### **CERCLA**

Not applicable

#### <u>CERCLA</u> SARA Product RQ

#### RCRA Pesticide Information

Component	FIFRA - Restricted Use	FIFRA - Pesticide Product Other Ingredients	FIFRA - Listing of Pesticide Chemicals	California Pesticides - Restricted Materials
Magnesium phosphide 12057-74-8 ( 56.0 (a.i.) )				Present

State Regulations

State Right-to-KnowNot applicableInternational regulationsU.S. EPA Label informationEPA Pesticide registration number70506-309

16. Other Information					
<u>NFPA</u>	<b>HEALTH</b> 4	flammability 2	Instability 0	Physical hazard -	
Preparation Date	04-Au	ıg-2015			
Revision date	31-De	ec-2018			
<b>Revision Summary</b>	/				
Update logo Update	e section 1 Update Section	n 16			
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